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Key indicators

Single-crystal X-ray study
 T = 100 K
 Mean $\sigma(C-C)$ = 0.001 Å
 R factor = 0.039
 wR factor = 0.103
 Data-to-parameter ratio = 23.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

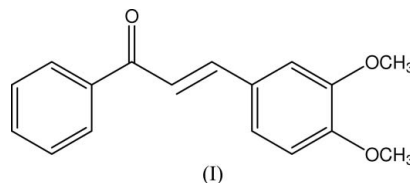
3,4-Dimethoxychalcone

In the title molecule, C₁₇H₁₆O₃, the dihedral angle between the benzene rings is 25.75 (3)°. The crystal packing is stabilized by C—H···π interactions.

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Comment

As part of our study on non-linear optical chalcone derivatives (Patil *et al.*, 2006*a,b,c*), we report here the synthesis and crystal structure of the title compound, (I). Crystals of (I) can potentially exhibit second-order non-linear optical properties as the compound crystallizes in a non-centrosymmetric space group. A quantitative estimation has yet to be performed.



Bond lengths and angles in (I) have normal values (Allen *et al.*, 1987), comparable with related structures (Teh *et al.*, 2006; Patil *et al.*, 2006*a,b,c*). The least-squares plane through the enone group (atoms C7–C9/O1) makes dihedral angles of 26.77 (5) and 25.32 (5)° with the C1–C6 and C10–C15 benzene rings, respectively. The dihedral angle between the two benzene rings is 25.75 (3)°. The methoxy groups at C12 and C13 are almost coplanar with the C10–C15 benzene ring, with C16–O2–C12–C11 and C17–O3–C13–C14 torsion angles of –6.29 (15) and –9.36 (14)°, respectively.

An intramolecular C9–H9A···O1 hydrogen bond generates an S(5) ring motif (Bernstein *et al.*, 1995). The crystal structure is stabilized by C—H···π interactions involving the C1–C6 benzene ring (Table 1).

Experimental

Acetophenone (0.01 mol) in ethanol (25 ml) was mixed with 3,4-dimethoxybenzaldehyde (0.01 mol) in ethanol (25 ml) and the mixture was treated with an aqueous solution of sodium hydroxide (5 ml, 30%). This mixture was stirred well and left for 24 h. The resulting crude solid mass was collected by filtration and recrystallized from acetone.

Crystal data

C ₁₇ H ₁₆ O ₃	Z = 16
M _r = 268.30	D _x = 1.330 Mg m ⁻³
Orthorhombic, Fdd2	Mo Kα radiation
a = 27.7541 (4) Å	μ = 0.09 mm ⁻¹
b = 34.1948 (4) Å	T = 100.0 (1) K
c = 5.6487 (1) Å	Block, yellow
V = 5360.88 (14) Å ³	0.51 × 0.37 × 0.33 mm

Data collection

Brucker SMART APEX2 CCD
area-detector diffractometer
 ω scans
Absorption correction: multi-scan
(SADABS; Bruker, 2005)
 $T_{\min} = 0.882$, $T_{\max} = 0.971$

41010 measured reflections
4202 independent reflections
3915 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$
 $\theta_{\text{max}} = 39.1^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.103$
 $S = 1.08$
4202 reflections
183 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0615P)^2 + 1.2638P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.43 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.19 \text{ e } \text{\AA}^{-3}$

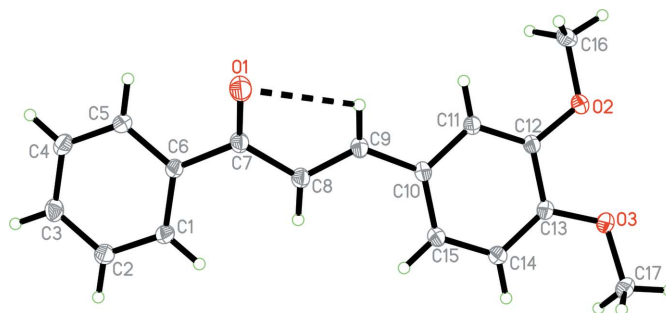


Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atomic numbering. The intramolecular hydrogen bond is shown as a dashed line.

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C9-H9A\cdots O1$	0.93	2.42	2.786 (1)	103
$C17-H17C\cdots Cg1^i$	0.96	2.75	3.384 (1)	124

Symmetry code: (i) $-x + 1, -y, z + 1$. Cg1 is the centroid of the C1–C6 ring.

H atoms were placed in calculated positions, with C–H = 0.93 or 0.96 \AA . U_{iso} values were set equal to $1.5U_{\text{eq}}$ of the carrier atom for methyl H atoms and $1.2U_{\text{eq}}$ for the remaining H atoms. In the absence of significant anomalous dispersion effects, Friedel pairs were merged.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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